

E0032

Dimensional Reduction and Recombination of the CdX₂ Lattice; New Extended Polyhedral Systems Based on CdX_n Organoammonium Species. B. Twamley[†], A. Thorn[‡], R. D. Willett[‡], [†]Univ. Research Office, Univ. of Idaho, Moscow, Idaho, 83844, [‡]Dept. of Chemistry, Washington State Univ., Pullman, WA, 99164.

The crystal engineering of novel hybrid layered complexes is important in the design of new materials with unusual structures and physical properties. Following our previous research,¹ we have investigated the use of organoammonium species as molecular scissors to modify the CdX₂ hexagonal lattice to obtain new extended 1D, 2D, and 3D systems. Ribbon structures, based on edge-shared CdX₆ and CdX₅L (L = H₂O or MeOH) octahedra, are obtained with stoichiometries [Cd₂Cl₅(H₂O)]_nⁿ⁻, [Cd₃Cl₈(MeOH)]_n²ⁿ⁻, [Cd₃Br₈(CdBr₂)]_n²ⁿ⁻, and [Cd₅Cl₁₂(H₂O)₂]_n²ⁿ⁻. In addition, a microporous version of the CdX₂ lattice, with stoichiometry [Cd₉Cl₂₀]_n²ⁿ⁻, is found using the (*n*-Bu)₂NH₂⁺ cation, while the use of the (Me)₂NH₂⁺ cation yields a novel 3D structure with stoichiometry [Cd₅Br₁₂]_n²ⁿ⁻, containing interlocked linear chains of CdBr₆ octahedra linked by isolated CdBr₄ tetrahedra.

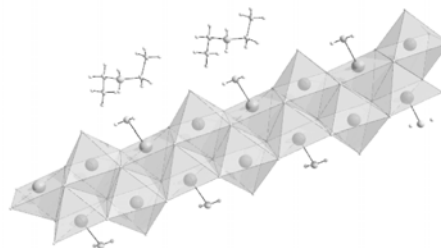


Figure 1. Ribbon structure of [(DEA)Cd₂Cl₅·H₂O]_n.

¹A. Thorn, R. D. Willett and B. Twamley, *Cryst. Growth & Des.* (ASAP article)